

# Free convection in a square porous cavity filled with a nanofluid using thermal non equilibrium and Buongiorno models

Free convection in a square porous cavity

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## Abstract

**Purpose** – The purpose of this paper is to theoretically analysis the steady-state natural convection flow and heat transfer of nanofluids in a square enclosure filled with a porous medium saturated with a nanofluid considering local thermal non-equilibrium (LTNE) effects. Different local temperatures for the solid phase of the nanoparticles, the solid phase of porous matrix and the liquid phase of the base fluid are taken into account.

**Design/methodology/approach** – The Buongiorno's model, incorporating the Brownian motion and thermophoresis effects, is utilized to take into account the migration of nanoparticles. Using appropriate non-dimensional variables, the governing equations are transformed into the non-dimensional form, and the finite element method is utilized to solve the governing equations.

**Findings** – The results show that the increase of buoyancy ratio parameter ( $N_r$ ) decreases the magnitude of average Nusselt number. The increase of the nanoparticles-fluid interface heat transfer parameter ( $N_{hp}$ ) increases the average Nusselt number for nanoparticles and decreases the average Nusselt number for the base fluid. The nanofluid and porous matrix with large values of modified thermal capacity ratios ( $\gamma_b$  and  $\gamma_s$ ) are of interest for heat transfer applications.

**Originality/value** – The three phases of nanoparticles, base fluid and the porous matrix are in the LTNE. The effect of mass transfer of nanoparticles due to the Brownian motion and thermophoresis effects are also taken into account.

**Keywords** Nanofluids, Free convection

**Paper type** Research paper

## Nomenclature

*Latin Symbols*

$C$  nanoparticle volume fraction

$C_0$  ambient nanoparticle volume fraction

$D_B$  Brownian diffusion coefficient



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$D_T$	thermophoretic diffusion coefficient	$T_h$	temperature at the left wall
$\mathbf{g}$	gravitational acceleration vector	$\mathbf{V}$	Darcy velocity
$h_{fp}$	interface heat transfer coefficient between the fluid/particle phases	$\bar{x}, \bar{y}$	Cartesian coordinates
$h_{fs}$	interface heat transfer coefficient between the fluid/solid-matrix phases	$\bar{u}, \bar{v}$	the velocity components along $\bar{x}, \bar{y}$ directions
$K$	permeability of the porous medium	<i>Greek symbols</i>	
$k$	effective thermal conductivity	$\alpha$	effective thermal diffusivity
$L$	cavity size	$\beta$	thermal expansion coefficient
$Le$	Lewis number	$\gamma_p$	modified particle heat capacity
$Nb$	Brownian motion parameter	$\gamma_s$	modified porous solid-matrix thermal conductivity
$Nh_p$	Nield number for the fluid/ nanoparticle interface (fluid/ nanoparticle interface parameter)	$\epsilon$	porosity
$Nh_s$	Nield number for the fluid/solid- matrix interface (fluid/solid-matrix interface parameter)	$\epsilon_p$	modified diffusivity ratio
$Nr$	buoyancy ratio parameter	$\theta$	dimensionless temperature
$Nt$	thermophoresis parameter	$\mu$	dynamic viscosity
$Nu$	local Nusselt number	$\rho$	fluid density
$\overline{Nu}$	average Nusselt number	$(\rho c)$	effective heat capacity
$p$	pressure	$\tau$	parameter defined by $\tau = (\rho c)_p / (\rho c)_f$
$Ra$	thermal Rayleigh–Darcy number $Ra = (1 - C_0)gK\rho_f\beta\Delta TL / (\alpha_f\mu)$	$\phi$	relative nanoparticle volume fraction
$Sh$	local Sherwood number	$\psi$	dimensionless stream function
$\overline{Sh}$	average Sherwood number	$\overline{\psi}$	dimensional stream function
$T$	dimensional temperature	<i>Subscripts</i>	
$T_c$	temperature at the right wall	0	the ambient property
		$f$	base fluid phase
		$p$	nanoparticle phase
		$s$	porous medium solid-matrix phase

### 1. Introduction

The free convection in enclosures filled with a porous medium saturated with a fluid occurs in many industrial applications including electronic component cooling, geothermal systems and paper production. In free convection flow there is no need of any external work source or fan and the fluid moves due to the temperature difference in the enclosure. Designing a system relying on a natural convection heat transfer reduces the risk of the mechanical system failure because of the absence of any external power supply. It also reduces the induced noise because of the fans. Because of such advantages many of engineers and researchers are interested in analysis of the natural convection phenomena and the systems working based on the natural convection heat transfer.

In many natural convection applications when the thermal conductivity of the porous medium is comparatively low or the interaction between the fluid and the porous medium is high, it can be assumed that the temperature of the fluid and the solid porous matrix are very close or they are in local thermal equilibrium (LTE). This assumption simplifies the problem in such a way that the fluid and the porous matrix can be seen as a uniform mixture. However, there are many practical cases of natural convection in porous media with high thermal removal capacities, in which there are significant temperature differences between the porous matrix and the fluid inside

the pores. In these cases, the LTE assumption is no longer valid, and the local thermal non-equilibrium (LTNE) model for the porous medium and the fluid should be taken into account (Ingham and Pop, 2005; Baytaş and Pop, 2002).

Recently, new-engineered fluids containing nanoparticles, namely, nanofluids, are developed to enhance the thermal conductivity of the working fluid (see Abu-Nada and Chamkha, 2010; Basak and Chamkha, 2012; Nasrin *et al.*, 2012, 2014; Chamkha and Abu-Nada, 2012). Experiments indicate that the presence of nanoparticles in the base fluid results in augmentation of the thermal conductivity. The increase in the thermal conductivity of a fluid increases the thermal diffusivity, and hence, it could result in heat transfer enhancement in convective heat transfer systems. Therefore, the nanofluids are potential candidates for convective heat transfer enhancement in applications with high demand of the thermal energy removal. There are also applications in which the presence of nanoparticles induces other desired effects beyond the heat transfer enhancement. For example, the presence of zinc oxide or titanium dioxide nanoparticles in water shows inherent antibacterial effects (Zhang *et al.*, 2008; Hirota *et al.*, 2010). The presence of ceramic oxide (Choi *et al.*, 2008) or diamond nanoparticles (Saidur *et al.*, 2011) enhance the dielectric properties of the base fluid, which is of interest in electrical power transformers.

The experiments indicate that the presence of nanoparticles in a fluid affects the other thermo-physical properties such as dynamic viscosity and heat capacity of the resulting fluid (Khanafer and Vafai, 2011). This is while the convective heat transfer of nanofluids is affected by the thermal conductivity, the dynamic viscosity and other thermo-physical properties as well as the fluid flow characteristics or mass transfer mechanisms. Therefore, the analysis of convective heat transfer of nanofluids, considering changes in thermo-physical properties and heat transfer mechanisms is highly demanded for engineering applications of nanofluids. Buongiorno (2006) analyzed the nanoparticles mass transfer in nanofluids using scale analysis and found that the Brownian motion and thermophoresis are two dominant particle transfer mechanisms in nanofluids. The Brownian motion force tends to uniform nanoparticles in the fluid. The thermophoresis force originates from the temperature gradient in the base fluid. When the size of a particle is very fine (in the order of nanometer), the particle receives more momentum impacts from the fluid molecules in hot side than that of the cold side; hence, the particle tends to move in a direction opposite to the temperature gradient (the particle moves from hot to cold). Therefore, there are mass transfer mechanisms in nanofluids because of the Brownian motion and thermophoresis forces. The migrated nanoparticles carry energy and transfer it to the surrounding medium. When the thermal conductivity of the base fluid is very low, the quiescent fluid around the nanoparticle cannot adequately absorb the heat from the nanoparticle, and hence, the nanoparticles and the base fluid could be in local thermal non-equilibrium.

The natural convection heat transfer of a regular fluid in an enclosure using LTE model has been well studied by previous fundamental researches. For example, Baytaş and Pop (1999, 2001) studied the natural convection of a regular fluid in enclosures using LTE model. Choi *et al.* (1998) have considered the non-Darcy effects in natural convection of regular fluids in enclosures filled with a porous medium. Varol *et al.* (2008) conducted a numerical study to analyze the effect of the shape of the enclosure on the natural convective heat transfer of nanofluids.

The LTNE natural convection of a regular fluid in an enclosure has been investigated by Baytaş and Pop (2002). The heat transfer of nanofluid in porous media

using LTE was formulated by Nield and Kuznetsov (2009a, b) considering the Buongiorno's (2006) mathematical model. They (Nield and Kuznetsov, 2009a) analyzed the heat and mass transfer of nanofluids over a flat plate embedded in a Darcy porous medium using a similarity solution approach. Nield and Kuznetsov (2009b) have utilized the Buongiorno's mathematical model to analysis the linear stability of nanofluids between two infinite plates using LTE model. Later, Bhadauria and Agarwal (2011) have developed the model of Nield and Kuznetsov (2009b) to study the natural convection instability of nanofluids between two infinity plates using the LTNE model. Very recently, Sheremet and Pop (2014, 2015) utilized the Buongiorno's model developed by Nield and Kuznetsov (2009a, b) to analysis the natural convection heat transfer of nanofluids in an enclosure.

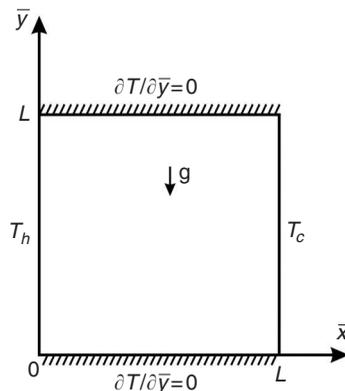
The present study aims to analysis the natural convection flow and heat transfer of nanofluids using the LTNE model considering the Buongiorno's model. In the present study, the three phases of nanoparticles, base fluid and the porous matrix are in the LTNE. The effect of mass transfer of nanoparticles due to the Brownian motion and thermophoresis effects are also taken into account.

## 2. Basic equations

Consider the steady free convection in a two-dimensional porous square cavity filled with a water-based nanofluid. It is assumed that nanoparticles are suspended in the nanofluid using either surfactant or surface charge technology. This prevents nanoparticles from agglomeration and deposition on the porous matrix (Kuznetsov and Nield, 2010, 2013; Nield and Kuznetsov, 2009a, 2014). A schematic view of the geometry of the problem under investigation is shown in Figure 1, where  $\bar{x}$  and  $\bar{y}$  are the Cartesian coordinates and  $L$  is the size of the cavity.

It is assumed that the left vertical wall is heated and maintained at the constant temperature  $T_h$ , while the right vertical wall is cooled and has the constant temperature  $T_c$ . The horizontal walls are adiabatic ( $\partial T / \partial \bar{y} = 0$ ), where  $T$  is the temperature. The Darcy-Boussinesq approximation and the LTNE between the fluid and solid-matrix and fluid and particle phase have been considered, thus heat flow has been described using three temperature model.

The conservation equations for the total mass, momentum, thermal energy in the fluid phase, thermal energy in the particle phase, thermal energy in the solid-matrix phase, and nanoparticles, come out to be as below. A detailed derivation of these



**Figure 1.**  
Physical model and  
coordinate system

equations has been dealt by Buongiorno (2006), Tzou (2008a, b), Nield and Kuznetsov (2009a), and Kuznetsov and Nield (2010):

$$\nabla \cdot \mathbf{V} = 0 \quad (1)$$

$$\frac{\mu}{K} \mathbf{V} = -\nabla p + [C\rho_p + (1-C)\rho_{f0}(1-\beta(T_f - T_c))] \mathbf{g} \quad (2)$$

$$\begin{aligned} \frac{1}{\varepsilon} \mathbf{V} \cdot \nabla T_f = & \frac{k_f}{(\rho c)_f} \nabla^2 T_f + \tau \left( D_B \nabla C \cdot \nabla T_f + \frac{D_T}{T_c} \nabla T_f \cdot \nabla T_f \right) \\ & + \frac{[h_{fp}(T_p - T_f) + h_{fs}(T_s - T_f)]}{\varepsilon(1-C_0)(\rho c)_f} \end{aligned} \quad (3)$$

$$\frac{1}{\varepsilon} \mathbf{V} \cdot \nabla T_p = \frac{k_p}{(\rho c)_p} \nabla^2 T_p + \frac{h_{fp}}{\varepsilon C_0 (\rho c)_p} (T_f - T_p) \quad (4)$$

$$0 = \frac{k_s}{(\rho c)_s} \nabla^2 T_s + \frac{h_{fs}}{(1-\varepsilon)(\rho c)_s} (T_f - T_s) \quad (5)$$

$$\frac{1}{\varepsilon} \mathbf{V} \cdot \nabla C = D_B \nabla^2 C + \frac{D_T}{T_c} \nabla^2 T_f \quad (6)$$

In these equations, both Brownian transport and thermophoresis coefficients are taken to be time-independent, in tune with the recent studies that neglect the effect of thermal transport attributed to the small size of the nanoparticles (as per recent arguments by Koblinski and Cahill, 2005). Further, thermophoresis and Brownian transport coefficients are assumed to be temperature-independent due to the fact that the temperature ranges under consideration are not far away from the critical value, and the volume averages over a representative elementary volume.

Equations (1)-(6) for the problem under consideration can be written in dimensional Cartesian coordinates  $\bar{x}, \bar{y}$  taking into account the slow flow and dilute nanoparticle concentration as:

$$\frac{\partial \bar{u}}{\partial \bar{x}} + \frac{\partial \bar{v}}{\partial \bar{y}} = 0 \quad (7)$$

$$\frac{\partial \bar{p}}{\partial \bar{x}} = -\frac{\mu}{K} \bar{u} \quad (8)$$

$$\frac{\partial \bar{p}}{\partial \bar{y}} = -\frac{\mu}{K} \bar{v} - [C(\rho_p - \rho_{f0}) + \rho_{f0}(1-\beta(T_f - T_c)(1-C_0))] g \quad (9)$$

$$\begin{aligned} \frac{1}{\varepsilon} \left( \bar{u} \frac{\partial T_f}{\partial \bar{x}} + \bar{v} \frac{\partial T_f}{\partial \bar{y}} \right) = & \alpha_f \left( \frac{\partial^2 T_f}{\partial \bar{x}^2} + \frac{\partial^2 T_f}{\partial \bar{y}^2} \right) + \tau \left\{ D_B \left( \frac{\partial C}{\partial \bar{x}} \frac{\partial T_f}{\partial \bar{x}} + \frac{\partial C}{\partial \bar{y}} \frac{\partial T_f}{\partial \bar{y}} \right) \right. \\ & \left. + \left( \frac{D_T}{T_c} \right) \left[ \left( \frac{\partial T_f}{\partial \bar{x}} \right)^2 + \left( \frac{\partial T_f}{\partial \bar{y}} \right)^2 \right] \right\} \end{aligned}$$

$$+ \frac{[h_{fp}(T_p - T_f) + h_{fs}(T_s - T_f)]}{\varepsilon(1 - C_0)(\rho c)_f} \tag{10}$$

$$\frac{1}{\varepsilon} \left( \bar{u} \frac{\partial T_p}{\partial \bar{x}} + \bar{v} \frac{\partial T_p}{\partial \bar{y}} \right) = \alpha_p \left( \frac{\partial^2 T_p}{\partial \bar{x}^2} + \frac{\partial^2 T_p}{\partial \bar{y}^2} \right) + \frac{h_{fp}}{\varepsilon C_0 (\rho c)_p} (T_f - T_p) \tag{11}$$

$$0 = \alpha_s \left( \frac{\partial^2 T_s}{\partial \bar{x}^2} + \frac{\partial^2 T_s}{\partial \bar{y}^2} \right) + \frac{h_{fs}}{(1 - \varepsilon)(\rho c)_s} (T_f - T_s) \tag{12}$$

$$\frac{1}{\varepsilon} \left( \bar{u} \frac{\partial C}{\partial \bar{x}} + \bar{v} \frac{\partial C}{\partial \bar{y}} \right) = D_B \left( \frac{\partial^2 C}{\partial \bar{x}^2} + \frac{\partial^2 C}{\partial \bar{y}^2} \right) + \left( \frac{D_T}{T_c} \right) \left( \frac{\partial^2 T_f}{\partial \bar{x}^2} + \frac{\partial^2 T_f}{\partial \bar{y}^2} \right) \tag{13}$$

where  $\bar{u}, \bar{v}$  are the velocity components along  $\bar{x}, \bar{y}$  directions, respectively.  
Introducing a stream function  $\bar{\psi}$  defined by:

$$\bar{u} = \frac{\partial \bar{\psi}}{\partial \bar{y}}, \quad \bar{v} = -\frac{\partial \bar{\psi}}{\partial \bar{x}}, \tag{14}$$

so that Equation (7) is satisfied identically. We are then left with the following equations:

$$\frac{\partial^2 \bar{\psi}}{\partial \bar{x}^2} + \frac{\partial^2 \bar{\psi}}{\partial \bar{y}^2} = -\frac{(1 - C_0)\rho_{f0}gK\beta}{\mu} \frac{\partial T_f}{\partial \bar{x}} + \frac{\rho_p - \rho_{f0}}{\mu} gK \frac{\partial C}{\partial \bar{x}}, \tag{15}$$

$$\begin{aligned} \frac{1}{\varepsilon} \left( \frac{\partial \bar{\psi}}{\partial \bar{y}} \frac{\partial T_f}{\partial \bar{x}} - \frac{\partial \bar{\psi}}{\partial \bar{x}} \frac{\partial T_f}{\partial \bar{y}} \right) &= \alpha_f \left( \frac{\partial^2 T_f}{\partial \bar{x}^2} + \frac{\partial^2 T_f}{\partial \bar{y}^2} \right) + \tau \left\{ D_B \left( \frac{\partial C}{\partial \bar{x}} \frac{\partial T_f}{\partial \bar{x}} + \frac{\partial C}{\partial \bar{y}} \frac{\partial T_f}{\partial \bar{y}} \right) \right. \\ &\quad \left. + \left( \frac{D_T}{T_c} \right) \left[ \left( \frac{\partial T_f}{\partial \bar{x}} \right)^2 + \left( \frac{\partial T_f}{\partial \bar{y}} \right)^2 \right] \right\} \\ &\quad + \frac{[h_{fp}(T_p - T_f) + h_{fs}(T_s - T_f)]}{\varepsilon(1 - C_0)(\rho c)_f} \end{aligned} \tag{16}$$

$$\frac{1}{\varepsilon} \left( \frac{\partial \bar{\psi}}{\partial \bar{y}} \frac{\partial T_p}{\partial \bar{x}} - \frac{\partial \bar{\psi}}{\partial \bar{x}} \frac{\partial T_p}{\partial \bar{y}} \right) = \alpha_p \left( \frac{\partial^2 T_p}{\partial \bar{x}^2} + \frac{\partial^2 T_p}{\partial \bar{y}^2} \right) + \frac{h_{fp}}{\varepsilon C_0 (\rho c)_p} (T_f - T_p) \tag{17}$$

$$0 = \alpha_s \left( \frac{\partial^2 T_s}{\partial \bar{x}^2} + \frac{\partial^2 T_s}{\partial \bar{y}^2} \right) + \frac{h_{fs}}{(1 - \varepsilon)(\rho c)_s} (T_f - T_s) \tag{18}$$

$$\frac{1}{\varepsilon} \left( \frac{\partial \bar{\psi}}{\partial \bar{y}} \frac{\partial C}{\partial \bar{x}} - \frac{\partial \bar{\psi}}{\partial \bar{x}} \frac{\partial C}{\partial \bar{y}} \right) = D_B \left( \frac{\partial^2 C}{\partial \bar{x}^2} + \frac{\partial^2 C}{\partial \bar{y}^2} \right) + \left( \frac{D_T}{T_c} \right) \left( \frac{\partial^2 T_f}{\partial \bar{x}^2} + \frac{\partial^2 T_f}{\partial \bar{y}^2} \right) \tag{19}$$

Introducing the following dimensionless variables:

$$\begin{aligned} x &= \bar{x}/L, \quad y = \bar{y}/L, \quad \psi = \bar{\psi}/\alpha_f, \quad \phi = C/C_0, \\ \theta_f &= (T_f - T_c)/\Delta T, \quad \theta_p = (T_p - T_c)/\Delta T, \quad \theta_s = (T_s - T_c)/\Delta T \end{aligned} \tag{20}$$

where  $\Delta T = T_h - T_c$ , and substituting (20) into Equations (15)-(19), we obtain:

$$\frac{\partial^2 \psi}{\partial x^2} + \frac{\partial^2 \psi}{\partial y^2} = -Ra \frac{\partial \theta_f}{\partial x} + Ra \cdot Nr \frac{\partial \phi}{\partial x} \quad (21)$$

$$\frac{\partial \psi}{\partial y} \frac{\partial \theta_f}{\partial x} - \frac{\partial \psi}{\partial x} \frac{\partial \theta_f}{\partial y} = \varepsilon \left( \frac{\partial^2 \theta_f}{\partial x^2} + \frac{\partial^2 \theta_f}{\partial y^2} \right)$$

$$+ Nb \left( \frac{\partial \phi}{\partial x} \frac{\partial \theta_f}{\partial x} + \frac{\partial \phi}{\partial y} \frac{\partial \theta_f}{\partial y} \right) + Nt \left[ \left( \frac{\partial \theta_f}{\partial x} \right)^2 + \left( \frac{\partial \theta_f}{\partial y} \right)^2 \right]$$

$$+ Nhp (\theta_p - \theta_f) + Nhs (\theta_s - \theta_f) \quad (22)$$

$$\frac{\partial \psi}{\partial y} \frac{\partial \theta_p}{\partial x} - \frac{\partial \psi}{\partial x} \frac{\partial \theta_p}{\partial y} = \varepsilon_p \left( \frac{\partial^2 \theta_p}{\partial x^2} + \frac{\partial^2 \theta_p}{\partial y^2} \right) + Nhp \cdot \gamma_p (\theta_f - \theta_p) \quad (23)$$

$$0 = \frac{\partial^2 \theta_s}{\partial x^2} + \frac{\partial^2 \theta_s}{\partial y^2} + Nhs \cdot \gamma_s (\theta_f - \theta_s) \quad (24)$$

$$\frac{\partial \psi}{\partial y} \frac{\partial \phi}{\partial x} - \frac{\partial \psi}{\partial x} \frac{\partial \phi}{\partial y} = \frac{1}{Le} \left( \frac{\partial^2 \phi}{\partial x^2} + \frac{\partial^2 \phi}{\partial y^2} \right) + \frac{Nt}{Le \cdot Nb} \left( \frac{\partial^2 \theta_f}{\partial x^2} + \frac{\partial^2 \theta_f}{\partial y^2} \right) \quad (25)$$

The corresponding boundary conditions for these equations are given by:

$$\psi = 0, \quad \theta_f = \theta_p = \theta_s = 1, \quad Nb \frac{\partial \phi}{\partial x} + Nt \frac{\partial \theta_f}{\partial x} = 0 \quad \text{on } x = 0$$

$$\psi = 0, \quad \theta_f = \theta_p = \theta_s = 0, \quad Nb \frac{\partial \phi}{\partial x} + Nt \frac{\partial \theta_f}{\partial x} = 0 \quad \text{on } x = 1$$

$$\psi = 0, \quad \frac{\partial \theta_f}{\partial y} = \frac{\partial \theta_p}{\partial y} = \frac{\partial \theta_s}{\partial y} = 0, \quad \frac{\partial \phi}{\partial y} = 0 \quad \text{on } y = 0 \text{ and } y = 1 \quad (26)$$

here the nine parameters  $Nr$ ,  $Nb$ ,  $Nt$ ,  $Nhp$ ,  $Nhs$ ,  $\varepsilon_p$ ,  $\gamma_p$ ,  $\gamma_s$ , and  $Le$  denote a buoyancy ratio parameter, a Brownian motion parameter, a thermophoresis parameter, the interface heat transfer parameters called as Nield numbers (Vadász, 2008), a modified thermal diffusivity ratio, modified thermal capacity ratios and Lewis number, respectively, which are defined as:

$$Nr = \frac{(\rho_p - \rho_{f0})C_0}{\rho_{f0}\beta\Delta T(1-C_0)}, \quad Nb = \frac{\tau D_B C_0 \varepsilon}{\alpha_f}, \quad Nt = \frac{\tau D_T \varepsilon \Delta T}{\alpha_f T_c}, \quad Nhp = \frac{h_{fp} L^2}{k_f(1-C_0)},$$

$$Nhs = \frac{h_{fs} L^2}{k_f(1-C_0)}, \quad \varepsilon_p = \frac{\alpha_p \varepsilon}{\alpha_f}, \quad \gamma_p = \frac{(1-C_0)(\rho c)_f}{C_0(\rho c)_p}, \quad \gamma_s = \frac{k_f(1-C_0)}{k_s(1-\varepsilon)}, \quad Le = \frac{\alpha_f}{D_B \varepsilon} \quad (27)$$

The physical quantities of interest are the local Nusselt numbers  $Nu_f$ ,  $Nu_p$ ,  $Nu_s$ , the local Sherwood number  $Sh$  and the average Nusselt numbers  $\overline{Nu}_f$ ,  $\overline{Nu}_p$ ,  $\overline{Nu}_s$  and Sherwood number  $\overline{Sh}$ .

The local Nusselt and Sherwood numbers are defined as:

$$Nu_f = -\left(\frac{\partial\theta_f}{\partial x}\right)_{x=0}, \quad Nu_p = -\left(\frac{\partial\theta_p}{\partial x}\right)_{x=0},$$

$$Nu_s = -\left(\frac{\partial\theta_s}{\partial x}\right)_{x=0}, \quad Sh = -\left(\frac{\partial\phi}{\partial x}\right)_{x=0} \quad (28)$$

The average Nusselt and Sherwood numbers are defined as:

$$\overline{Nu}_f = \int_0^1 Nu_f dy, \quad \overline{Nu}_p = \int_0^1 Nu_p dy,$$

$$\overline{Nu}_s = \int_0^1 Nu_s dy, \quad \overline{Sh} = \int_0^1 Sh dy \quad (29)$$

It should be noted here that for an analysis of Sherwood numbers it is possible to study only Nusselt numbers because at the left and right vertical walls we have  $\partial\phi/\partial x = -(Nt/Nb)(\partial\theta_f/\partial x)$  taking into account boundary conditions for  $\phi$  (Equation (26)). Therefore the further analysis concerning integral parameters will be about only average Nusselt number because  $Sh = -(Nt/Nb)Nu_f$  and  $\overline{Sh} = -(Nt/Nb)\overline{Nu}_f$ .

In the present study the average Nusselt number for the base fluid phase is reported as average  $Nu_f = \varepsilon\overline{Nu}_f$  and for the nanoparticles phase as average  $Nu_p = \varepsilon_p\overline{Nu}_p$  to explicitly show the non-dimensional heat transfer rare of the corresponding phase.

### 3. Numerical method and validation

The set of partial differential Equations (21)-(25) subject to the boundary conditions (26) were transformed to weak form and solved using the finite element method (Rao, 2005; Wriggers, 2008). The Lagrange shape function and quadratic elements were utilized in the finite element method (Rao, 2005). The governing equations were fully coupled using damped Newton method (Wriggers, 2008). The solution for the corresponding linear algebraic equations was obtained using a parallel sparse direct solver (Amestoy *et al.*, 2000). The computations are terminated when the residuals for the stream function get bellow  $10^{-8}$ . The present model, in the form of an in-house computational fluid dynamics code, has been validated successfully against the works of Mansour *et al.* (2010), Chamkha and Ismael (2013), Costa (2004), Chamkha *et al.* (2010) and Baytaş and Pop (2002) for steady-state natural convection in an enclosure filled with porous media.

Considering the previous studies, the value of  $Nb$  and  $Nt$  are small in the order of  $10^{-6}$ . The Lewis number is large of the order of  $10^3$  and higher, and  $Nr$  is higher than unity. The interface heat transfer parameters, Nield numbers ( $Nhp$  and  $Nhs$ ), are higher than unity (Bhadauria and Agarwal, 2011). The modified thermal diffusivity ratio ( $\varepsilon_p$ ) is about unity and higher, and the modified thermal capacity ratios ( $\gamma_p$  and  $\gamma_s$ ) are order of 10. Here, the results are present for the following typical combination of non-dimensional parameters  $Ra = 100$ ;  $Nr = 10.0$ ;  $Nb = 10^{-6}$ ;  $Nt = 10^{-6}$ ;  $Le = 1,000$ ;  $\varepsilon = 0.5$ ;  $\varepsilon_p = 1.0$ ;  $Nhp = 10.0$ ;  $Nhs = 10.0$ ;  $\gamma_p = 10.0$  and  $\gamma_s = 10.0$ .

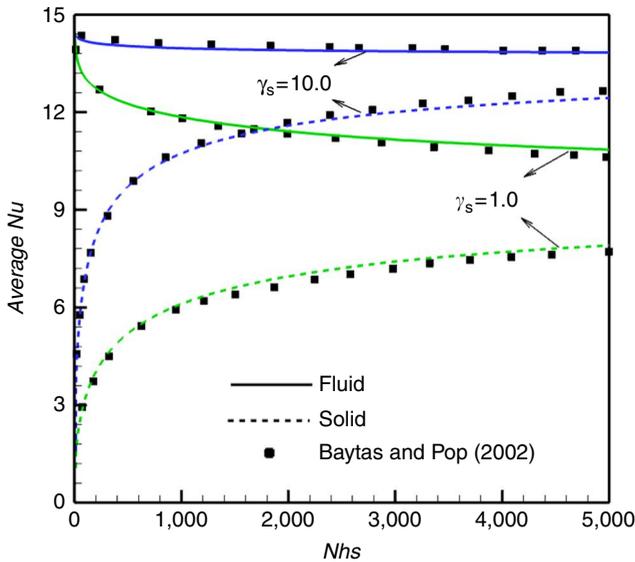
Table I shows the evaluated Nusselt numbers of various phases for different combinations of the dimensionless interface heat transfer parameters and different grid sizes. Table I shows that the grid size of  $150 \times 150$  provides accurate results. Hence, all of the calculations were executed using the grid size of  $150 \times 150$ .

Neglecting the effect of nanoparticles ( $Nr = Nb = Nt = Nhp = 0$ ), the present study reduces to the study of LTNE in a porous medium saturated with a pure fluid which was analyzed in the work of Baytaş and Pop (2002). In this case, dividing both sides of Equation (22) by  $\epsilon$ , the non-dimensional form of the present study is equivalent with the study of Baytaş and Pop (2002). In this case, a comparison between the computed values of the average Nusselt number and those reported by Baytaş and Pop (2002) is performed in Figure 2. Table II also shows the evaluated values of the average Nusselt

Grid size	$Nhp = Nhs = 0$			$Nhp = Nhs = 10$		
	$Nu_f$	$Nu_s$	$Nu_p$	$Nu_f$	$Nu_s$	$Nu_p$
50 × 50	2.4733	1.0000	2.7015	2.3083	2.3105	3.3942
100 × 100	2.4723	1.0000	2.7001	2.3071	2.3120	3.3938
150 × 150	2.4720	1.00000	2.6998	2.3069	2.3122	3.3937
200 × 200	2.4719	1.00000	2.6997	2.3068	2.3123	3.3936

**Notes:**  $Ra = 100$ ;  $Nr = 10.0$ ;  $Nb = 10^{-6}$ ;  $Nt = 10^{-6}$ ;  $Le = 1,000$ ;  $\epsilon = 0.5$ ;  $\epsilon_p = 1.0$ ;  $\gamma_p = 10.0$  and  $\gamma_s = 10.0$

**Table I.**  
Grid independency test



**Figure 2.**  
Comparison with the results of Baytaş and Pop (2002) for  $\epsilon = 0.9$  and neglecting nanoparticles

$Nhs$	Baytaş and Pop (2002)						Present study					
	$\gamma_s = 1$		$\gamma_s = 10$		$\gamma_s = 50$		$\gamma_s = 1$		$\gamma_s = 10$		$\gamma_s = 50$	
	$Nu_f$	$Nu_s$	$Nu_f$	$Nu_s$	$Nu_f$	$Nu_s$	$Nu_f$	$Nu_s$	$Nu_f$	$Nu_s$	$Nu_f$	$Nu_s$
5	14.36	1.26	14.44	2.59	14.64	4.63	14.21	1.26	14.38	2.62	14.46	4.69
10	14.14	1.48	14.51	3.35	14.63	5.76	14.00	1.48	14.34	3.39	14.45	5.83
50	13.43	2.51	14.4	5.70	14.6	8.69	13.36	2.54	14.24	5.77	14.42	8.74
100	13.1	3.18	14.34	6.89	14.58	9.95	13.06	3.23	14.18	6.97	14.41	9.94
500	12.24	5.11	14.16	9.71	14.54	12.42	12.25	5.22	14.03	9.73	14.38	12.20

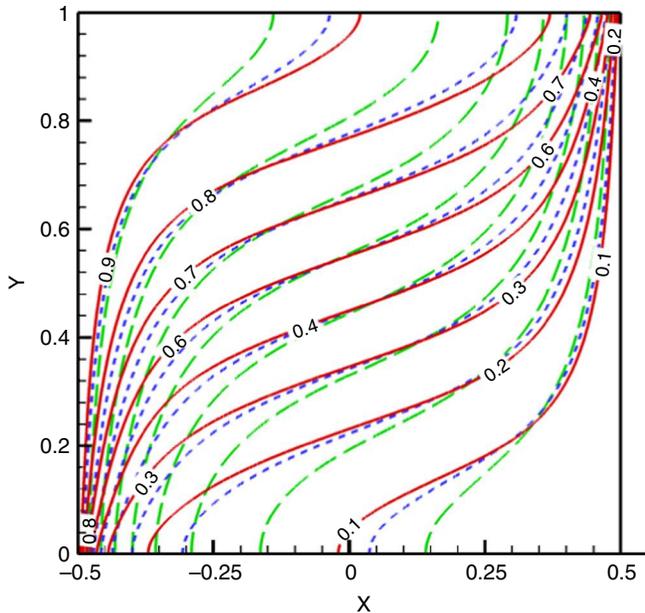
**Notes:**  $Ra = 1,000$ ;  $Nr = Nb = Nt = Nhp = 0$

**Table II.**  
Comparison of the average Nusselt number of the hot wall using LTNE model

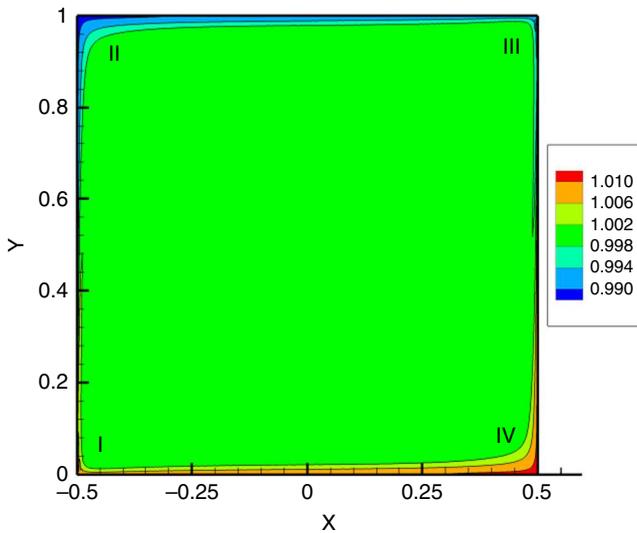
number of the hot wall for  $Ra = 1000$  and compares the results with the data reported by Baytaş and Pop (2002). As seen, there is a good agreement between the results of the present study and the results of Baytaş and Pop (2002).

#### 4. Results and discussion

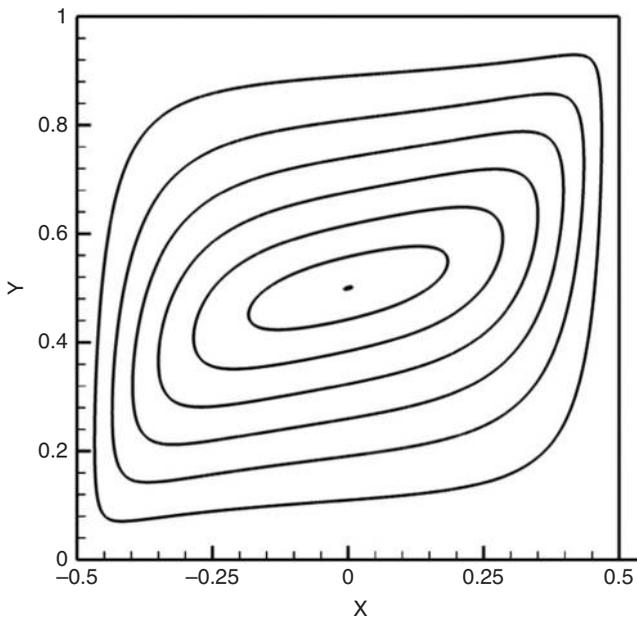
Figures 3-5 show the contours of the isotherms, concentration of nanoparticles and the nanofluid streamlines for a typical case of LTNE heat natural convection of nanofluids. As seen, there is a significance difference between isotherms of the base fluid, solid matrix and nanoparticles. This difference between the isotherms of three phases is small near isothermal walls where the buoyancy effects and fluid flow are strong. In contrast, the observed difference, between the isotherms of various phases, becomes essential one in the vicinity of the adiabatic horizontal walls where the fluid flow is weak. Figure 4 indicates that the concentration of nanoparticles near the hot vertical wall is low due to the thermophoresis effect. The thermophoresis force acts on the nanoparticles in direction opposite to the temperature gradient, which tends to move nanoparticles from hot to cold. As seen, the concentration of nanoparticles near the cold wall is high. In addition, the weakest concentration of nanoparticles can be seen in the upper left corner of the enclosure (II). Attention to Figure 3 indicates that the temperature gradient near the bottom left corner (I) of the enclosure is higher than that of the corner (II) as the isotherms are denser close to the corner (I). Hence, in the vicinity of the left wall (hot wall), moving from the bottom to top would decrease the strength of the temperature gradients. As the thermophoresis is a direct function of temperature gradient, the intensity of the thermophoresis force would also follow the strength of the temperature gradient. When the nanofluid starts to move upward from the bottom of the hot wall, the thermophoresis tends to push away the nanoparticles in direction away from the hot wall. Hence, the concentration of nanoparticles tends to decrease



**Figure 3.** Isotherms inside the cavity: the solid red lines correspond to the base fluid  $\theta_b$ ; the long dashed green lines correspond to the porous matrix  $\theta_s$  and the small dashed blue lines correspond to the nanoparticles  $\theta_p$



**Figure 4.** Contours of the nanoparticles concentrations inside the enclosure

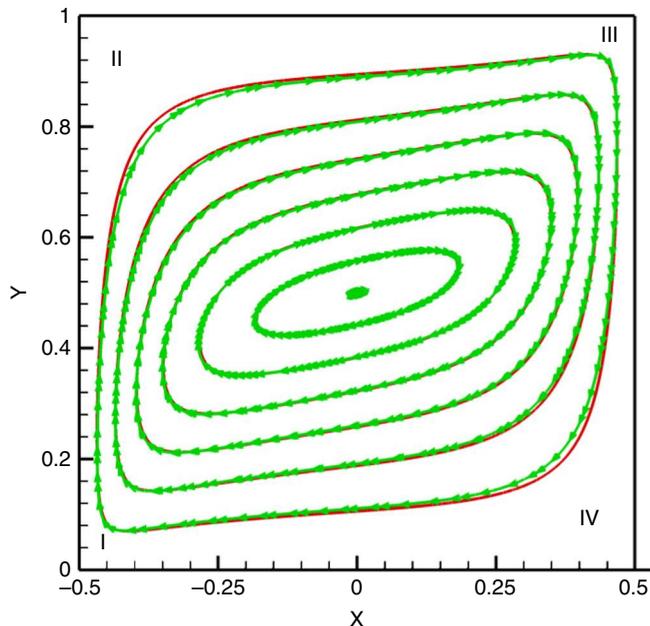


**Figure 5.** Streamlines of the nanofluid inside the cavity

towards upward. Then, the flow of nanofluid moves along the horizontal adiabatic top wall until it reaches to the cold wall at right. Thus, the weakest concentration of nanoparticles can be seen in the corner (II) and near the top horizontal adiabatic wall. When the flow of the nanofluid, containing weak concentration of nanoparticles, reaches the cold wall, the temperature gradients get strong again and induce

thermophoresis forces toward the cold wall. Hence, the concentration of nanoparticles increases next to the cold wall, and the strongest concentration of nanoparticles can be seen at the bottom right corner (IV). Figure 4 also shows that the concentration boundary layer of nanoparticles is a thin layer, and the concentration in most parts of the enclosure's core is uniform. The thin concentration boundary layer of nanoparticles is due to the value of Lewis number in nanofluids. The Lewis number shows the ratio of the thermal diffusivity to the nanoparticles diffusion coefficient in nanofluids. For nanofluids, the thermal diffusivity is much higher than the nanoparticles diffusion coefficient, and hence, the Lewis number is a large value. In the theory of the boundary layer, the Lewis number indicates the comparative ratio of the thickness of the thermal boundary layer to the thickness of the concentration boundary layer. The thickness of the thermal boundary layer is mostly affected by the hydrodynamic boundary layer and the buoyancy forces; hence, the thickness of the thermal boundary layer is almost fixed. Therefore, as the consequent of the large values of the Lewis number, the concentration boundary layer is very thin. Figure 5 shows the streamlines of the nanofluid in the enclosure. As seen, the streamlines illustrate the flow circulation of nanofluid inside the cavity. The streamlines are close together near the isothermal walls where the temperature gradients and buoyancy forces are strong. Hence, the velocity near the vertical walls is high and near the core is low.

Figure 6 shows the effect of the buoyancy ratio parameter ( $Nr$ ) on the isotherms. When nanoparticles migrate from a place into another one, the areas containing high volume fraction of nanoparticles are heavier than that of containing weak concentration of nanoparticles. Hence, the concentration difference would induce the buoyancy force because of the nanoparticles mass transfer. The buoyancy ratio parameter indicates the ratio of buoyancy force due to the mass transfer of nanoparticles to the buoyancy force due to the heat transfer. The increase of  $Nr$

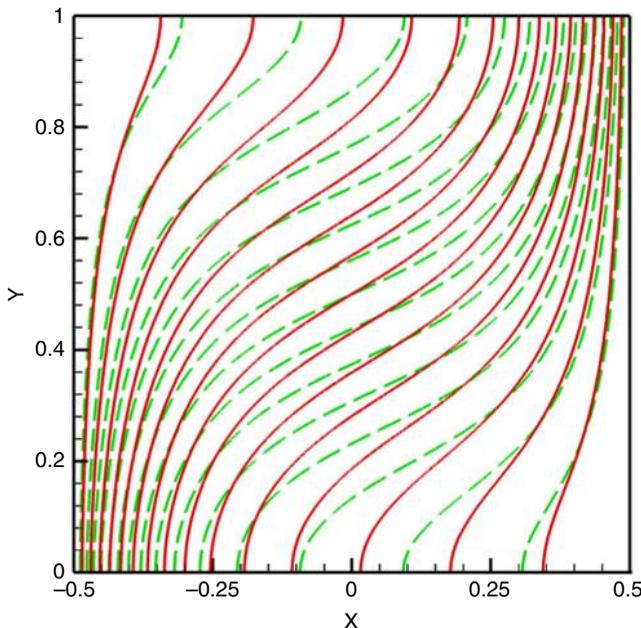


**Figure 6.**  
The streamlines for the nanofluid for  $Nr=0$  (green lines marked with arrows) and  $Nr=30$  (clean red lines)

intensifies the effect of induced buoyancy force due to the mass transfer (migration of nanoparticles). Thus, as seen in Figure 6, the effect of buoyancy ratio parameter ( $Nr$ ) only near the corners of (II) and (IV) is significant. These places correspond to the weakest and strongest concentration of nanoparticle in the enclosure. The difference between the streamlines in the core of the enclosure, where the mass transfer mechanism are negligible, is also negligible.

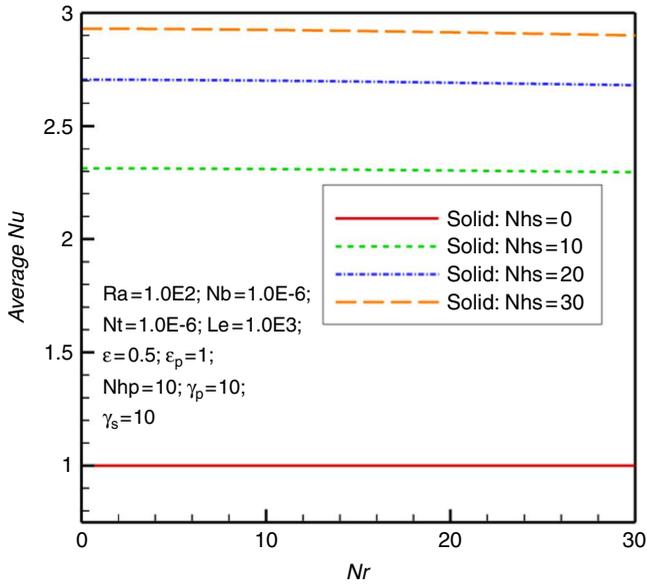
Figure 7 shows the effect of  $Nhs$  on the isotherms of the porous matrix.  $Nhs$  is the interface heat transfer parameter between the fluid phase and the solid phase. Decreasing  $Nhs$ , the effect of fluid phase on the solid matrix reduces, and hence, for low values of  $Nhs$  the solid matrix would act as an isolated solid interface in which the temperature would change linearly from hot to cold. Figure 7 also indicates that the isotherms would better follow the non-linear behaviors of the fluid for high values of  $Nhs$  (dashed green lines). It is also clear that the variation of the interface heat transfer parameter between the fluid phase and the solid phase would significantly affect the isotherms. The effect of  $Nhs$  on the concentration and fluid phase isotherms was not significant, and hence, they have not been plotted here for the sake of brevity.

Figures 8-10 show the average Nusselt number of the solid, base fluid and nanoparticles phases, respectively. In these figures, the effect of the buoyancy ratio parameter on the average Nusselt numbers for selected values of the solid-fluid interface heat transfer parameter ( $Nhs$ ) is illustrated. As seen, the increase of the buoyancy ratio parameter ( $Nr$ ) leads to the decrease in the magnitude of the average Nusselt number. The observed decrease in the base fluid (Figure 9) and nanoparticles (Figure 10) phases is more significant in comparison with that of the porous solid matrix (Figure 8). The increase of the buoyancy ratio parameter intensifies the induced buoyancy mass transfer effects. As it was seen in Figure 6, these effects are solely important in a small layer close to the vertical isothermal walls. Therefore, the increase

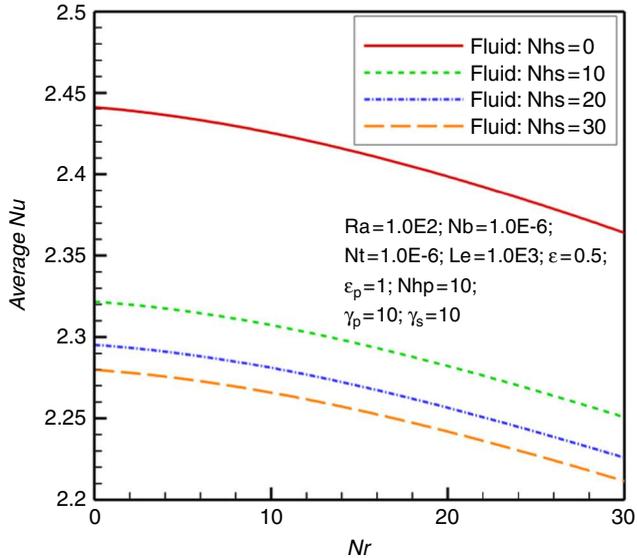


**Figure 7.** Isotherms for the porous matrix temperature  $\theta_s$  for  $Nhs = 5$  (solid red lines) and  $Nhs = 10$  (the long dashed green lines)

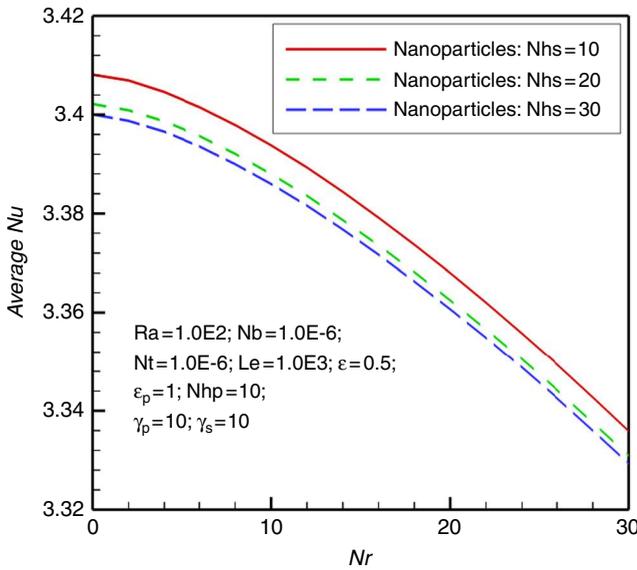
**Figure 8.** Average Nusselt number of the porous matrix as a function of the buoyancy ratio parameter ( $Nr$ ) for various values of the solid-fluid interface heat transfer parameter ( $Nhs$ )



**Figure 9.** Average Nusselt number of the fluid as a function of the buoyancy ratio parameter ( $Nr$ ) for different values of the solid-fluid interface heat transfer parameter ( $Nhs$ )



of the buoyancy ratio parameter boosts the vertical velocity of the fluid in the vicinity of the vertical walls, which results in the decrease of temperature gradient (average Nusselt number). The variation of the buoyancy ratio parameter induces direct significant changes in the buoyancy forces next to the vertical walls which results in



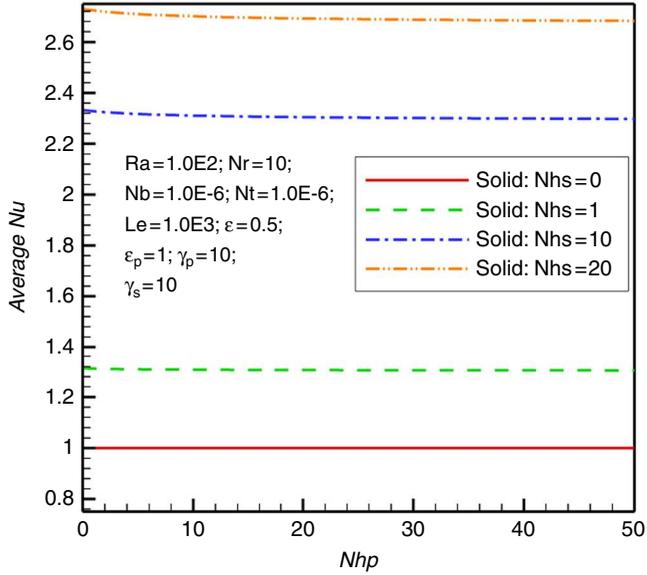
**Figure 10.** Average Nusselt number of the nanoparticles as a function of the buoyancy ratio parameter ( $Nr$ ) for various values of the solid-fluid interface heat transfer parameter ( $Nhs$ )

smooth indirect variation of average Nusselt number (temperature gradient) of the base fluid. The variation of the base fluid would consequently affect the temperature gradient of the solid porous medium. However, as seen in Figure 8, this indirect effect is not much significant. Figure 9 indicates that the increase of the solid-fluid interface heat transfer parameter decreases the average Nusselt number for the base fluid phase. In contrast, Figure 8 indicates that the increase of  $Nhs$  increases the average Nusselt number for the phase of the solid porous matrix. This is because of the fact that the increase of  $Nhs$  boosts the interaction between the fluid and solid matrix, and consequently tends to reduce the temperature differences between these two phases.

The changes in temperature behavior of nanoparticles are mainly the results of the interaction between nanoparticles and the base fluid. Therefore, the temperature behavior of nanoparticles tends to follow the temperature behavior of the base fluid through interaction with the base fluid. Thus, as seen in Figure 10, the increase of  $Nhs$  decreases the average Nusselt number for the nanoparticles phase. This is the same trend of behavior as occurred for the base fluid, but the variation of Nusselt number for the nanoparticles phase is smoother than that for the base fluid. This is because of the fact that the variation of  $Nhs$  directly affects the temperature of the solid porous matrix and the base fluid phases, but the fluctuations of  $Nhs$  would indirectly change the temperature behavior of the nanoparticles through the variation of base fluid temperature.

Figures 11-13 illustrate the effect of the nanoparticles-fluid interface heat transfer parameter ( $Nhp$ ) on the average Nusselt number of the three phases of the solid porous matrix, base fluid and nanoparticles for various values of the solid-fluid interface heat transfer parameter ( $Nhs$ ). The nanoparticles-fluid interface heat transfer parameter ( $Nhp$ ) indicates the strength of thermal interaction between nanoparticles and the base fluid. As this parameter increases, the interaction between these phases increases.

**Figure 11.** Average Nusselt number of the porous matrix as a function of the nanoparticles-fluid interface heat transfer parameter ( $Nhp$ ) for various values of solid-fluid interface heat transfer parameter ( $Nhs$ )



**Figure 12.** Average Nusselt number of the fluid phase as a function of the nanoparticles-fluid interface heat transfer parameter ( $Nhp$ ) for various values of solid-fluid interface heat transfer parameter ( $Nhs$ )

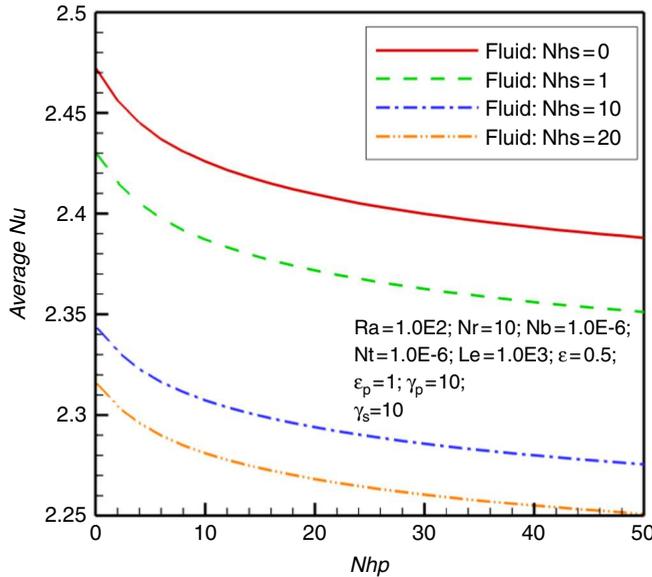
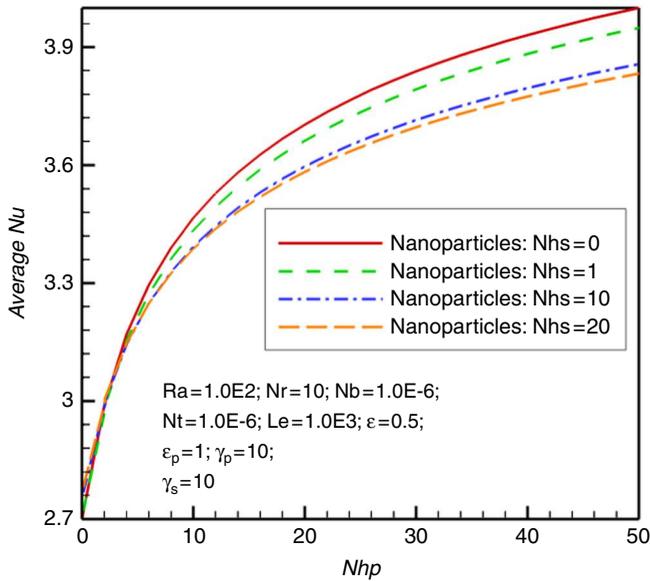


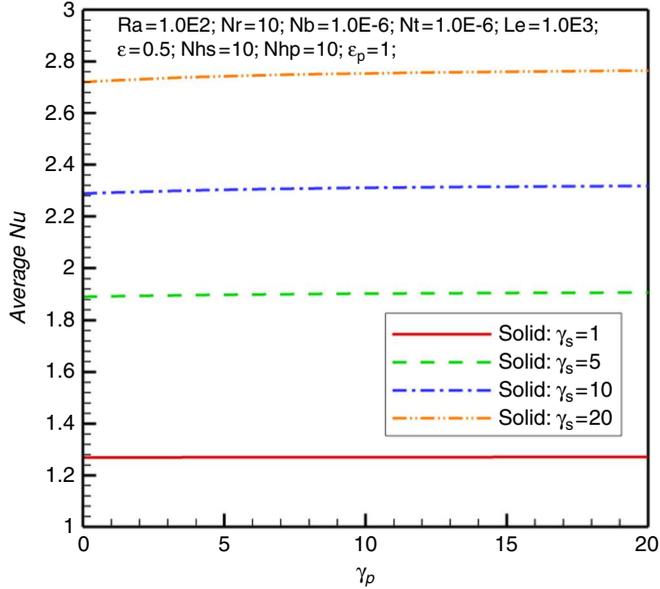
Figure 11 reveals that the effect of variation of  $Nhp$  on the average Nusselt number is not significant for the solid phase of the porous matrix. This is because the fluctuations of  $Nhp$  would indirectly affect the temperature behavior of the solid porous phase through the variation of base fluid temperature. This figure in agreement with Figure 8 shows that the increase of  $Nhs$  increases the average Nusselt number of the solid porous matrix.



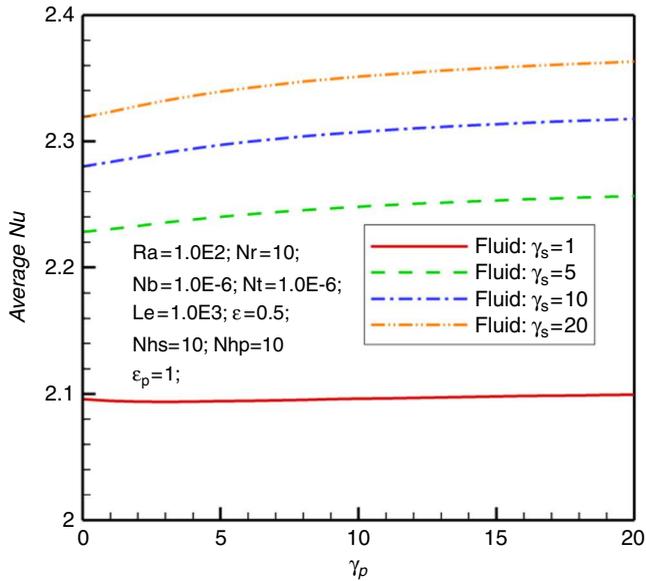
**Figure 13.** Average Nusselt number of the nanoparticles phase as a function of the nanoparticles-fluid interface heat transfer parameter ( $Nh_p$ ) for various values of solid-fluid interface heat transfer parameter ( $Nh_s$ )

Figures 12 and 13 depict that the increase of  $Nh_p$  tends to reduce the temperature gradient (average Nusselt number) for the base fluid phase and raise the average Nusselt number for the nanoparticles. This is because of the fact that the increase of the interaction between base fluid and nanoparticles tends to reduce the temperature differences between the phases. However, as the heat capacity and volume fraction of nanoparticles are lower than those of the base fluid, the changes in temperature gradient (i.e. reduced Nusselt number) in both sides is not equal. In addition, as it was seen in previous figures, the variation of the solid-fluid interface heat transfer parameter ( $Nh_s$ ) also plays an important role in surface temperature gradients of the phases (i.e. Nusselt numbers).

Figures 14-16 illustrate the average Nusselt number of three phases of the solid porous matrix, base fluid and nanoparticles as a function of modified thermal capacity ratio of nanoparticle phase ( $\gamma_p$ ) for various values of the modified thermal capacity ratio of the solid matrix of the porous medium phase ( $\gamma_s$ ). These figures depict that the increase of the modified thermal capacity ratio of nanoparticle phase ( $\gamma_p$ ) increases the average Nusselt number of all phases when the modified thermal capacity ratio of the porous solid matrix ( $\gamma_s$ ) is comparatively high. The increase of  $\gamma_s$  would also increase the average Nusselt number for the solid porous matrix and the base fluid, but it induces two different trends of results for the nanoparticle phase. When  $\gamma_p$  is comparatively small ( $\gamma_p < 2$ ), the increase of  $\gamma_s$  decreases the average Nusselt number of nanoparticles, but when  $\gamma_p$  is comparatively large, the increase of  $\gamma_s$  increases the average Nusselt number for the nanoparticles phase. Attention to the governing Equations (21)-(25) indicates that variation of  $\gamma_p$  can reduce or boost the effect of  $Nh_p$  in the heat transfer of nanoparticles phase. Similarly, the variation of  $\gamma_s$  can reduce or intensify the effect of  $Nh_s$  in the phase of the porous solid matrix. Hence, when the magnitude of  $\gamma_s$  is large the temperature of the solid matrix is strongly under influence of the variation of the base fluid phase. In addition, when the magnitude of  $\gamma_p$  is high the

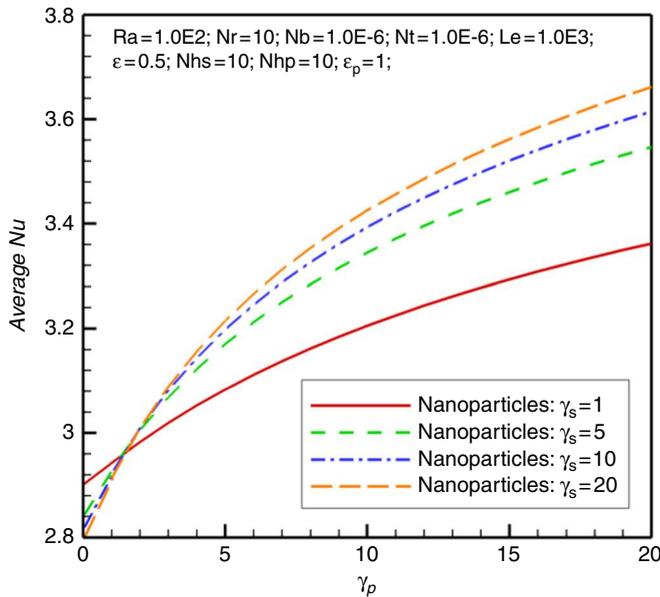


**Figure 14.** Average Nusselt number of the porous matrix as a function of  $\gamma_p$  for various values of  $\gamma_s$



**Figure 15.** Average Nusselt number of the fluid phase as a function of  $\gamma_p$  for various values of  $\gamma_s$

temperature of nanoparticles is under the strong influence of the base fluid phase. Therefore, in the case of the large values of  $\gamma_p$  and  $\gamma_s$ , the temperatures (as well as the Nusselt numbers) should follow the behavior of the base fluid. Figure 15 reveals that the variation of  $\gamma_p$  can smoothly change the magnitude of the average Nusselt number of the base fluid. However, it induces strong effects on the magnitude of the Nusselt



**Figure 16.** Average Nusselt number of the fluid phase as a function of  $\gamma_p$  for various values of  $\gamma_s$

number of the nanoparticles phase. Figure 14 indicates that the variation of  $\gamma_p$  does not show significant effects on the variation of the Nusselt number of the solid porous matrix. This is because of the fact that the variation of  $\gamma_p$  would indirectly change the temperature gradients of the base fluid through the variation of nanoparticles temperatures. Then, the smooth variation of the base fluid temperature would change the temperatures of the solid matrix of the porous medium. Figure 14 shows that the variation of  $\gamma_s$  induces significant changes in the magnitude of the average Nusselt number of the solid-matrix phase. As the magnitude of  $\gamma_s$  increases, the effect of interaction between solid-base fluid phases (i.e.  $Nhs$ ) intensifies in the solid porous matrix side. In Figure 8, it was observed that the increase of  $Nhs$  tends to increase the average Nusselt number of the solid matrix of porous medium. Here, the increase of  $\gamma_s$  ( $\gamma_s = 10$ ) has intensified this effect and increased the reduced Nusselt number.

As mentioned, Figure 15 shows that the increase of  $\gamma_s$  increases the reduced Nusselt number for the base fluid. Indeed, the increase of  $\gamma_s$  indirectly affects the Nusselt number of the base fluid through the variation of the temperature of the solid porous matrix. When  $\gamma_p$  is large any changes in temperature of the fluid would be boosted for nanoparticles, and the temperature of the base fluid dominantly changes the temperature of nanoparticles. Hence, as seen in Figure 16, the increase of  $\gamma_s$  raises the magnitude of the average Nusselt number for large values of  $\gamma_p$ . Indeed, this trend of results follows the trend of results of Figure 15. In contrast, when  $\gamma_p$  is small, the variation of the temperature of the base fluid could be neglected, and the changes in temperature of nanoparticles would be the results of the diffusive term as well as the velocity of the base fluid. It is clear that the diffusive term is not under influence of  $\gamma_s$ . Therefore, the changes in the average Nusselt number of nanoparticles correspond to velocity (advection) terms for the case of small values of  $\gamma_p$ . Figure 15 indicates as  $\gamma_s$

increases the average Nusselt number for the base fluid (Figure 15), and consequently, the velocity increases. The increment of the velocity reduces the temperature gradient of nanoparticles. Hence, as seen in Figure 16, the increase of  $\gamma_s$  reduces the magnitude of the average Nusselt number for small values of  $\gamma_p$ .

## 5. Conclusion

The natural convective flow and heat transfer of nanofluids in a square enclosure filled with a homogenous porous medium saturated with a nanofluid is theoretically analyzed. The nanofluid and porous medium are considered in LTNE. The nanoparticles in the nanofluid are also subject to drift flux due to the Brownian motion and thermophoresis forces. The governing equations in non-dimensional form were solved using a finite element code. The results for the heat transfer of nanofluids were reported in the form of the Nusselt number graphs for three phases of solid porous matrix, base fluid and nanoparticles. The main outcomes of the results of the present study can be summarized as follows:

- (1) The increase of buoyancy ratio parameter ( $Nr$ ) decreases the magnitude of average Nusselt number.
- (2) The increase of the nanoparticles-fluid interface heat transfer parameter ( $Nhp$ ) increases the average Nusselt number for nanoparticles and decreases the average Nusselt number for the base fluid. Similarly, the increase of the solid-fluid interface heat transfer parameter ( $Nhs$ ) increases the average Nusselt number for the porous matrix and the nanoparticles and decreases the average Nusselt number for the base fluid. The effect of variation of  $Nhp$  on the average Nusselt number of the porous matrix is not significant.
- (3) The increase of the modified thermal capacity ratio of nanoparticle phase ( $\gamma_p$ ) increases the average Nusselt number of all phases when the modified thermal capacity ratio of the porous solid matrix ( $\gamma_s$ ) is comparatively high. Therefore, nanofluids with large values of modified thermal capacity ratio are of interest for heat transfer applications.
- (4) The increase of the modified thermal capacity ratio of the porous solid matrix ( $\gamma_s$ ) raises the average Nusselt number for all phases when the modified thermal capacity ratio nanoparticle phase ( $\gamma_p$ ) is comparatively high. In practical cases, the values of  $\gamma_p$  are large, and hence, in heat transfer applications, the mixtures of nanofluids and porous matrix with high values of  $\gamma_s$  are of interest.

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Free  
convection in  
a square  
porous cavity

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